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$\text{Ca}_3(\text{PO}_4)_2(\text{s}) + 2\text{H}_2\text{SO}_4(\text{aq})$
 $\rightarrow 3\text{CaSO}_4(\text{s}) + 2\text{H}_3\text{PO}_4(\text{aq})$

→ 4 + 3 4 q) Coefficient of H₂SO₄
(aq) Coefficient of H₃PO₄ (aq) A.
1 2 B. 2 3 C. 3 1 D. 3 2 3. 7.102 g
of 1 Na₂SO₄ (142.04g/mol) 2 4 M = –
is dissolved in water to prepare
0.5000 dm³ of solution. What is
the concentration of Na₂SO₄ in
mol dm⁻³? A. 2.500×10⁻² B.
1.000×10⁻¹ C. 1.000 ...

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