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Molarity of H₂SO₄

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Solution 0.05 M 0.5 M 0.05 M
0.05M Average Molarity 0.6M.

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Reaction 1: $1 \text{ NaHCO}_3 (\text{s}) + 1 \text{ NaOH} (\text{s}) + 1 \text{ CO}_2 (\text{g})$
Reaction 2: $2 \text{ NaHCO}_3 (\text{s}) \rightarrow 1 \text{ Na}_2\text{CO}_3 (\text{s}) + 1 \text{ CO}_2 (\text{g}) + 1 \text{ H}_2\text{O} (\text{g})$
Reaction 3: $2 \text{ NaHCO}_3 (\text{s}) \rightarrow 1 \text{ Na}_2\text{O} (\text{s}) + 2 \text{ CO}_2 (\text{g})$

Propose a voltaic cell that would coat a piece of copper with platinum. Include the half reactions for your cell and the voltage potential based on the.

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answer is your local wastewater treatment facility, which operates 24/7 to make sure

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Answers - Maharashtra

1. Identify the group for each unknown. Your unknowns will be from three different groups. (Example: A K^+ salt, a halogen, and an acid.) 2. Identify the specific compound from each group. (Example: KCl , Br_2 and HCl) 3. Explain the reasoning that enabled you to come to these conclusions.

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