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Chemical Equilibrium Lab Answers

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the equilibrium position refers to the relative amounts of reactants and products in the system at the point of equilibrium. 1:

Introduction to Equilibrium. mL/s (many other possible answer would work) 4.

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Balanced Chemical Reaction: Sign of

Chemical Reaction: Type of Chemical

Reaction: $KI(aq) + Ag(aq) \rightarrow KNO_3(aq) +$

$AgI(s)$ $KI(aq) + Ag(aq) \rightarrow KNO_3(aq) +$

$AgI(s)$ Change in colour: into an opaque

yellow. Liquid form. Solid cannot be seen:

Double Displacement: $CoCl_2(aq) +$

$Na_2SO_4(aq) \rightarrow CoSO_4(aq) + NaCl(aq)$

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