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CHEM201A Chapter 6 Lecture Outline Gottlieb 15 Example: Determine the mass of hydrogen gas, in g, that must be combusted to produce 1510.2 kJ of heat.  $2H_2(g) + O_2(g) \rightarrow 2H_2O(g)$   $\Delta H_{rxn} = -483.6 \text{ kJ}$   
Example: A 32.3-mL sample of 0.996M NaOH is mixed with 19.1 mL of 0.905M  $H_2SO_4$ .

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