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Chapter 12 Solutions review. liquid that dissolves a gas or solid or the solution component in greater amount. when a system in equilibrium is altered by a change of temperature, pressure, or concentration variable, the system shifts in equilibrium composition in a way that tends to counteract this change of variable.

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Chapter 12 Review: Solutions Name: Carly 1) Does a solution have to involve a liquid? Explain your answer. No; solutions may exist as gases, liquids, or solids. 2) Explain why a suspension is considered a heterogeneous mixture. The particles are so large that

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they settle out unless the mixture is constantly stirred or agitated.

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SHORT ANSWER Answer the following questions in the space provided. 1. Match the type of mixture on the left to its representative particle diameter on the right.

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c solutions (a) larger than 1000 nm a suspensions (b)
1 nm to 1000 nm b colloids (c)

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How does ...

Modern Chemistry Chapter 12 Solutions Review
Answers Modern Chemistry Chapter 12-Solutions
Section 1- Types of Mixtures Solutions are
homogeneous mixtures of two or more substances in a
single phase. Soluble describes a substance as capable

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of being dissolved. Solvent is the dissolving medium in a solution. Chapter 12 Review Solutions Section 3 Answers

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