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(continued) Write Lewis structures for  
each of the following molecules.  
Indicated the bonds with either dots of

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dashes. 15.  $\text{NH}_3$  18.  $\text{PCl}_3$ . 16.  $\text{H}_2$  19.  $\text{CCl}_4$ . 17.  $\text{C}_2\text{H}_4$  20.  $\text{C}_2\text{H}_6$ . Answer the following questions as directed. 21. Explain why the molecules  $\text{SF}_4$  is an exception to the octet rule.\

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each of the following molecules.

Indicated the bonds with either dots of dashes. 15.  $\text{NH}_3$  18.  $\text{PCl}_3$ . 16.  $\text{H}_2$  19.

$\text{CCl}_4$ . 17.  $\text{C}_2\text{H}_4$  20.  $\text{C}_2\text{H}_6$ . Answer the following questions as directed. 21.

Explain why the molecules  $\text{SF}_4$  is an exception to the octet rule.\ 7-2

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## Answers

Sunlevel 5-2 Review and Reinforcement (continued) Identify each of the following elements as a metal (M), nonmetal (NM), or semimetal (SM). M 19. sodium SM 20. silicon NM 21. neon M 22. calcium NM 23. nitrogen Write the family names

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that have been given to each of the following groups.

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