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protons? 2. How many protons and electrons are 9. What is the name of the element that has present in a nitrogen atom? atoms that contain 82 protons? 3. How many protons and electrons are 10.

This work evolved over thirty combined years of teaching general chemistry to a variety of student demographics. The focus is not to recap or review the theoretical concepts well described in the available texts. Instead, the topics and descriptions in this book make available specific, detailed step-by-step methods and procedures for solving the major types of problems

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in general chemistry. Explanations, instructional process sequences, solved examples and completely solved practice problems are greatly expanded, containing significantly more detail than can usually be devoted to in a comprehensive text. Many chapters also provide alternative viewpoints as an aid to understanding. Key Features: The authors have included every major topic in the first semester of general chemistry and most major topics from the second semester. Each is written in a specific and detailed step-by-step process for problem solving, whether mathematical or conceptual. Each topic has greatly expanded examples and solved practice problems containing significantly more detail than found

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Provides over 175 worked examples and more than 500 practice problems and quiz questions to help students develop and practice their problem solving skills.

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The Fill-in, Multiple Choice are two sets of questions that include every concept and numerical exercise introduced in the chapter and the Skills section provides developed exercises to apply the new concepts in the chapter to particular examples. The Common Mistakes section is designed to help avoid some of the errors that students make in their effort to learn chemistry, while the Practical Test section includes matching and multiple choice questions that comprehensively cover almost every concept and numerical problem in the chapter. After briefly dealing with an overview of chemistry, this book goes on exploring the concept of matter, energy, measurement, problem solving, atom, periodic table, and chemical

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bonding. These topics are followed by discussions on writing names and formulas of compounds; chemical formulas and the mole; chemical reactions; calculations based on equations; gases; and the properties of a liquid. The remaining chapters examine the solutions; acids; bases; salts; oxidation-reduction reactions; electrochemistry; chemical kinetics and equilibrium; and nuclear, organic, and biological chemistry. This study guide will be of great value to chemistry teachers and students.

The thoroughly revised & updated 9th Edition of Go To Objective NEET Chemistry is developed on the objective pattern following the chapter plan as per the



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news is that practice makes perfect, and this book provides plenty of it—with easy-to-understand coaching every step of the way. Delve deep into the parts of the periodic table Get comfortable with units, scientific notation, and chemical equations Work with states, phases, energy, and charges Master nomenclature, acids, bases, titrations, redox reactions, and more Understanding introductory chemistry is critical for your success in all science classes to follow; keeping up with the material now makes life much easier down the education road. Chemistry Workbook For Dummies gives you the practice you need to succeed!

The Practice of Medicinal Chemistry fills a gap in the

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list of available medicinal chemistry literature. It is a single-volume source on the practical aspects of medicinal chemistry. Considered "the Bible" by medicinal chemists, the book emphasizes the methods that chemists use to conduct their research and design new drug entities. It serves as a practical handbook about the drug discovery process, from conception of the molecules to drug production. The first part of the book covers the background of the subject matter, which includes the definition and history of medicinal chemistry, the measurement of biological activities, and the main phases of drug activity. The second part of the book presents the road to discovering a new lead compound and creating a working hypothesis. The main

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parts of the book discuss the optimization of the lead compound in terms of potency, selectivity, and safety. The Practice of Medicinal Chemistry can be considered a "first-read" or "bedside book" for readers who are embarking on a career in medicinal chemistry. NEW TO THIS EDITION: \* Focus on chemoinformatics and drug discovery \* Enhanced pedagogical features \* New chapters including: - Drug absorption and transport - Multi-target drugs \* Updates on hot new areas: NEW! Drug discovery and the latest techniques NEW! How potential drugs can move through the drug discovery/development phases more quickly NEW! Chemoinformatics

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